INDUCTION OF C-MYC G-QUADRUPLEX DNA AND CYTOTOXICITY OF A CALIX[4]ARENE-CONTAINING BINUCLEAR RUTHENIUM(II) COMPLEX

YA-XUAN MI^a, SHUANG WANG^a, SHUANG-MEI YAN^a, YUE CUI^a, MING-HE WANG^a, ZE-BAO ZHENG^{b*} AND XIAO-LONG ZHAO^{a*}

^aCollege of Chemistry & Environmental Science, Key Laboratory of Chemical biology of Hebei Province, Hebei University, Baoding 071002, P.R. China. ^bCollege of Chemistry and Chemical Engineering, Taishan University, Taian 271021, P.R. China.

ABSTRACT

Herein, a binuclear ruthenium(II) complex modified by calix[4]arene group has been prepared. The complex as potential inducer and stabilizer of *c-myc* Gquadruplex DNA and antitumor reagent were studied. Observations revealed that the complex could bind to *c-myc* Pu27 and Pu22 DNA strongly with constants of 1.18×10^7 (Pu27) and 4.13×10^6 M⁻¹ (Pu22) *via* groove mode as determined from absorption and luminescence titrations, as well as CD spectra. Results of continuous variation analysis confirmed that the complex interacted with *c-myc* G-quadruplex DNA through a 1:1 binding stoichiometry. As verified by PCR-stop assay, the replication of *c-myc* DNA was effectively blocked by the complex with the complete inhibition at the complex concentration of 4.0 μ M both for Pu27 and Pu22, suggesting that the complex could efficiently induce the formation of *c-myc* G-quadruplex DNA. The appearance of blue TMB solution in the visual experiment also proved that the sequences of Pu27 and Pu22 could fold into G-quadruplex under the induction of the complex. Nevertheless, the complex was found to exhibit weak stabilization ability on *c-myc* G-quadruplex DNA according to FRET assay, which increased the melting point of *c-myc* DNA only 3-3.5 °C. The experiments on Topoisomerase inhibition and cytotoxicity of the complex showed that it acted as an inhibitor of TopoI and exhibited moderate anticancer activity against MCF-7 and Huh-7 tumor cells.

Keywords: Ruthenium, c-myc G-quadruplex DNA, Topoisomerase, Cytotoxicity.

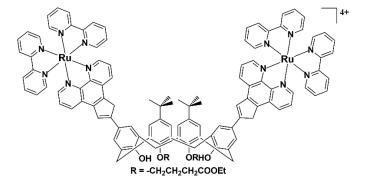
1. INTRODUCTION

Ruthenium(II) complexes have attracted extensively attentions as potential alternative drugs for *cis*-platin in metal-based cancer therapy, due to their excellent binding affinity and luminescene-probe behavior for DNA [1-4], high selectivity to tumor cells and relatively low toxicity toward human normal cells [5-8]. Research of the last few decades by Chao et al. revealed that binding of ruthenium(II) complexes with G-quadruplex DNA plays an important role in their inhibition of tumor cell growth [9-11]. It has been found that the G-quadruplex forming guanine-rich sequences widely exist in many regions of chromosomes [12], such as telomeres [13], promoters of oncogenes [14], immunoglobulin switches [15] and the insulin regulatory regions [16]. Particularly, the promoter of oncogene *c-myc*, which is overexpressed in up to 80% of solid tumors, but has low expression in normal cells, can also form a G-quadruplex conformation *via* Hoogesten hydrogen bonds, and is closely related to the proliferation, apoptosis, cell-cycle arrest, invasion and metastasis of tumor cells [17-19].

Therefore, the oncogene c-myc G-quadruplex DNA as a potential target for antitumor drugs has received increased attention. Up to now, a number of ruthenium(II) complexes have been reported to effectively bind to, induce and stabilize c-myc G-quadruplex DNA [11,20-27]. Zhang et al. found that complex $[Ru(phen)_2(p-tFMPIP)](ClO_4)_2$ {p-tFMPIP = 2-(4-(trifluoromethyphenyl)-1Himidazo[4,5f][1,10]phenanthroline)} can stabilize the conformation of c-myc Gquadruplex DNA in groove binding mode and inhibit the growth of MDA-MB-231 cells through apoptosis pathway [22]. Li et al. reported that the aryl alkyne modified complex [Ru(phen)₂(TMSEPIP)](ClO₄)₂ {TMSEPIP = 2-(2trimethylsilylethylphenyl)imidazole[4,5f][1,10]phenanthroline)} can insert into the groove of *c-myc* G-quadruplex DNA and can act as a potential luminescent switch-on probe through selectively recognizing and promoting self-assembly of c-myc G-quadruplex DNA [21]. Wu et al. demonstrated that a series of arene ruthenium(II) complexes $[(\eta^6-C_6H_6)Ru(p-XPIP)CI]CI (X = H; F; CI; Br; and I)$ {PIP = 2-phenylimidazole[4,5f][1,10]phenanthroline)} can bind and stabilize cmyc G-quadruplex DNA via groove mode and exhibit excellent inhibitory activity against MDA-MB-231 breast cancer cells [20].

However, among the ruthenium(II) complexes reported, limited investigations are focused on the binuclear complexes. Thomas and co-workers indicated that tppz (tetrapyrido[3,2-*a*:2',3'-c:3'',2''-*h*:2'',3'''-*j*]phenazine) and tatpp (tetraazatetrapyrido[3,2-*a*:2'3'-*c*:3'',2''-*l*:2''',3'''-*n*]pentacene) containing binuclear ruthenium(II) complexes could bind to G-quadruplex DNA with higher affinities than that of duplex DNA molecules **[28**].

Research results of Zheng et al. about [(bpy)2Ru(bpibp)Ru(bpy)2](ClO4)4 $\{bpibp = 4,4'-bis(1,10-phenanthroline-[5,6-d]imidazole-2-yl)-biphenyl\}$ and [(phen)₂Ru(bpibp)Ru(phen)₂](ClO₄)₄ evidenced that the binuclear complexes can stabilize the structure of G-quadruplex in the *c-myc* promoter and targeting the G-quadruplex loop isomers with inhibition of the proliferation of HepG2 cells [29]. It has been shown that the binuclear complexes generally possess increased size and charge, varied molecular shapes, DNA structural selectivity and great DNA binding affinity, compared to analogous mononuclear complexes, which are advantageous as DNA binders and structural probes [30]. This provokes us to design novel binuclear ruthenium (II) complexes and further understand their interaction with c-myc G-quadruplex DNA. With this in mind, a calix[4]arenebased binuclear ruthenium(II) complex was prepared in this paper (see Scheme 1), and the affinity of binding to and ability of induction, stabilization of *c-myc* G-quadruplex DNA, Topoisomerase inhibition and anticancer activity of the complex were investigated. The complex was proved to be a promising groove binder and inducer of c-myc Pu27 and Pu22 DNA, and identified as a TopoI inhibitor and potent antitumor agent against MCF-7 and Huh-7 cells.



Scheme 1. Molecular structure of ruthenium(II) complex.

2. EXPERIMENTAL

2.1 Materials

The calix[4]arene-based binuclear ruthenium(II) complex was prepared according to methods described in the literature [**31**]. The synthetic route and characterization of the complex were given in the supporting information. The molecular structure of the complex is shown in **Scheme 1**.

*Corresponding author email: longlong_666@sina.com, zhengzebao@163.com

The other chemicals were obtained from commercial sources and used as received. HTG21 (5'-GGGTTAGGGTTAGGGTTAGGGTTAGGG-3'), *c-myc* Pu27 (5'-TGGGGAGGGTGGGGAGGGTGGGGAAGG-3'), Pu27rev (5'-ATCGATC GCTTCTCGTCCTTCCCCA-3'), *c-myc* Pu22 (5'-GAGGGTGGGGAGGGTGG GGAAG-3'), Pu22rev (5'-ATCGCTTCTCGTCTTCCCCA-3') and the fluorescent labeled oligonucleotides F27T (5'-FAM-TGGGGAGGGTG GGGAGGGTGGGGAAGG-TAMRA-3', FAM: carboxyfluorescein, TAMRA: 6-carboxytetramethylrhodamine) and F22T (5'-FAM-GAGGGTGGGGAAGG-TAMRA-3') were purchased from Sangon Biotech (Shanghai) Co., Ltd. The G-quadruplex conformation was formed by denaturation at 90 °C for 5 min followed by renaturation at 4 °C for 24 h [**12,21,32**]. Several buffers of A: 10 mM Tris-HCl, pH = 7.4; B:10 mM Tris-HCl, 100 mM KCl, pH = 7.4; C: 10 mM Tris-HCl, 60 mM KCl, pH = 7.4 were applied to the experiments. All aqueous solutions were prepared with Ultrapure MilliQ water (18.2 MΩ).

2.2 DNA binding experiments

Electronic spectra were measured on a UV-3600 Shimadzu spectrophotometer. Emission spectra were done with a Shimadzu RF-5301PC spectrofluorophotometer. The absorption and luminescence titrations of the ruthenium complex in buffer A were performed with a fixed complex concentration to which increments of *c-myc* G-quadruplex DNA stock solutions were added sequentially. The complex–DNA solutions were incubated for 5 min before spectra were recorded. The titration processes were repeated until there was no change in the spectra, indicating binding saturation had been achieved.

Circular dichroism (CD) spectra were performed with a Mos-450/SFM300 spectropolarimeter. During the CD titration procedure, aliquots of ruthenium complex solution were added continuously to solutions of *c-myc* G-quadruplex DNA in buffer B. The sample solutions were mixed thoroughly and allowed to equilibrate for 5 min before data collection. For each sample, the spectrum was scanned at least three times and accumulated over the wavelength range of 230–320 nm. The titration was repeated until no further change was observed, indicating that the binding saturation had been achieved.

For continuous variation analysis that was conducted with the Shimadzu RF-5301PC spectrofluorophotometer, two series of solutions were used: one with varying mole fractions of [Ru]/[DNA], and another one with varying concentrations of complexes. The concentration of the Ru–DNA solutions was always 10 μ M. The emission spectra were collected from 500 to 750 nm. The ΔI values were calculated by subtracting the fluorescence intensity of complex solution without Pu27/Pu22 from the fluorescence intensity of corresponding complex solution with Pu27/Pu22 at λ_{max} . This value was plotted versus the complex mole fraction to generate a Job plot.

2.3 Induction of c-myc G-quadruplex DNA by ruthenium complex

The oligonucleotide Pu27/Pu22 and the corresponding complementary sequence Pu27rev/Pu22rev were used in the Polymerase chain reaction (PCR) stop assay. The PCR-stop assay was performed in $1 \times$ PCR buffer, containing 20 pmol of each oligonucleotide, 0.16 mM dNTPs, 2.5 U Taq polymerase, and different concentrations of ruthenium complex. Reaction mixtures were incubated in a thermocycler with the following cycling conditions: 94 °C for 3 min, followed by 30 cycles of 94 °C for 30 s, 58 °C for 30 s, and 72 °C for 30 s, then 72 °C for 10 min.

The visual detection of *c-myc* G-quadruplex structure was conducted as followes: An equal volume of ruthenium complex solution was added to DNA solutions (20 mM DNA, 10 mM Tris-HCl, 100 μ M EDTA, pH = 8.00), allowing the DNA strands to form the G-quadruplex structure in 40 min. An equal volume of hemin (in DMSO) was dissolved in the G-quadruplex solutions, which were then kept for 2 h at room temperature to form the DNAzymes. Subsequently, 180 μ L of 296 μ M TMB (3,3',5,5'-tetramethylbenzidine)-1.76 mM H₂O₂ solution was added as a substrate for the above 20 μ L peroxidatic DNAzyme system. The mixture was kept for 1.5 h at room temperature, and then different colors were observed with the naked eye.

2.4 Stabilization of *c-myc* G-quadruplex DNA by ruthenium complex

Fluorescence resonance energy transfer (FRET) melting point assay was carried out on a Roche real time PCR cycler. The fluorescent labeled oligonucleotide F27T/F22T used as the FRET probe was diluted in buffer C and

then annealed by heating to 92 °C for 5 min, followed by overnight cooling to room temperature. Duplex ct-DNA was a competitive binder to evaluate the selective binding ability of ruthenium complex with G-quadruplex DNA. Fluorescence melting curves were monitored by using a total reaction volume of 25 μ L, with labeled oligonucleotide (1 mM) and different concentrations of ruthenium complex in buffer C. Fluorescence readings with excitation at 470 nm and detection at 513 nm were taken at intervals of 1 °C from 37 to 95 °C.

2.5 Topoisomerase inhibition assay

DNA Topoisomerase I (Topo I) was purchased from New England Biolabs. The reaction mixture (15 μ L) used in the experiment contained 20 mM Tris-HCl, 50 mM KCl, 10 mM MgCl₂, 0.1 mg/mL BSA, 1 mM DTT (dithiothreitol), 0.25 μ g of pBR322 DNA and 2 Unit of Topo I along with varying concentrations of complex. The resultant reaction mixtures were incubated at 37 °C for 30 min, and stopped by adding 2 μ L of 6 × loading buffer. The electrophoresis of the samples was passed through 1% agarose gel (Biowest) in 1 × TBE buffer at 90 V for 1.5 h. The gel was stained with 1 mg/mL EB and visualized by UV light, and then photographed on a Bio-Rad gel imaging system.

2.6 MTT assay

The cytotoxicity of the ruthenium complex was investigated against A549 (human lung cancer cell), Huh-7 (human hepatocarcinoma cell), MGC-803 (human gastric cancer cell), MCF-7 (human breast cancer cell) and 293T (a normal human cell, human embryonic kidney cell) cell lines on the basis of MTT (3-(4,5-dimethylthiazole)-2,5-diphenyltetrazolium bromide) assay. Cisplatin was used as a reference chemotherapeutic drug. Cells were seeded into a 96-well plate and incubated under 5% CO₂ at 37 °C for 24 h in 100 µL of cell suspension at a density of 5×10^3 cells per well. The cells were then treated with various concentrations of the tested complex for 48 h. Upon completion of the incubation, stock MTT dye solution (20 µL) was added to each well. The microplate was incubated at 37 °C for another 4 h in the incubator, and finally sodium dodecyl sulfate (SDS, 100 µL) was added to dissolve the formed formazan crystals. The absorbance of formazan solution was read on a microplate reader at a wavelength of 570 nm.

3. RESULTS AND DISCUSSION

3.1 Binding behavior of ruth enium complex with c-myc G-quadruplex DNA

Absorption spectroscopy is one of the approved methods for examining the interaction of small molecules with biology macromolecules. The interaction of ruthenium complex with *c-myc* G-quadruplex has been firstly investigated by the electronic spectra, as shown in Fig. 1. The absorption spectra of the complex is characterized by a metal-to-ligand charge transfer (MLCT) absorption in the visible region (468 nm), and an intense $\pi \rightarrow \pi^*$ intra-ligand (IL) charge transfer in the ultraviolet region (288 nm). On addition of not only *c-myc* Pu27 DNA but also *c-myc* Pu22 DNA, obvious hypochromism and red shift were observed. The hypochromism at IL absorption reached as large as 50.6% with a red shift of 4 nm at Pu27 binding saturation and 48.4% accompanied by a bathochromic shift of 4 nm at Pu22 binding equilibration, although the changes of absorbance of the MLCT bands were slight when [*c-myc* G-quadruplex DNA] increased. These results gave a hint that the complex exhibited promising affinity to *c-myc* G-quadruplex DNA.

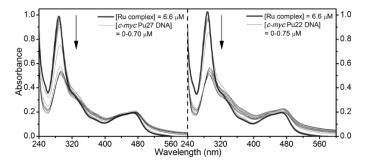


Figure 1. Electronic spectra of ruthenium complex in the absence and presence of *c-myc* G-quadruplex DNA.

To further clarify the ability of the complex binding to c-myc G-quadruplex DNA, a luminescence titration study was conducted. As illustrated in Fig. 2, excitation of the complex at 460 nm in Tris-KCl buffer resulted in an intense ³MLCT {Ru(d π)→ligand(π *)} emission band with λ_{max} around 610 nm. As increasing the concentration of c-myc Pu27 or Pu22 DNA, the situation of emission for the complex is interesting. Initially, the addition of c-myc DNA causes large reductions in fluorescence intensity, but after reaching a minimum, further additions lead to increasing luminescence intensity until it is higher than that observed before the addition of any c-myc DNA. The decrease of emission intensity displays that the complex binds to *c-mvc* DNA through groove mode. which would result in a quenching effect [29,33]. On the other hand, the hydrophobic environment inside the DNA leads to the increase of the emission intensity at the later stage of titration. This phenomenon is similar to that observed for [(phen)₂Ru(bpibp)Ru(phen)₂](ClO₄)₄ which has been shown to be groove binding to c-myc G-quadruplex DNA [29]. The DNA binding affinity of the complex can be evaluated quantitatively by monitoring the increasing of the emission dates in luminescence titration according to Bard-Torp-Murphy equations [34]. As shown in Fig. 2, the Pu27 and Pu22 binding constants were estimated to be 1.18×10^7 and 4.13×10^6 M⁻¹, respectively, which are much larger than 0.43, 0.77, 53.0, 51.0, and 4.5×10^5 M⁻¹ that reported for [(η^6 -C₆H₆)Ru(p-XPIP)Cl]Cl [20], demonstrating that the complex binds to c-myc DNA with higher strength.

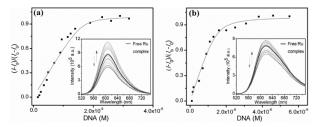


Figure 2. Emission spectra of ruthenium complex in the presence of increasing amounts of *c-myc* G-quadruplex DNA, and plot of $(I-I_0)/(I_b-I_0)$ vs. [DNA] and the nonlinear fitting. (a) *c-myc* Pu27 DNA, [Ru] = 3.75 μ M, [DNA] = 0-3.2 μ M, (b) *c-myc* Pu22 DNA, [Ru] = 3.75 μ M, [DNA] = 0-5.5 μ M.

CD spectra can also offer some useful information on the interaction of small molecules with biology macromolecules. After interaction with small molecules, the configuration of DNA will change accordingly and thus reveal different binding modes of the molecules to DNA. The CD spectra of c-myc G-quadruplex DNA at increasing concentrations of ruthenium complex in Tris-KCl buffer were illustrated in Fig. 3. The resulting spectra showed that c-myc G-quadruplex DNA exhibited a parallel structural conformation, which was characterized by a strong positive signal in the range of 250-300 nm with the maximum at 265 nm, in addition to a weak negative CD signal between 200 and 250 nm with the maximum at 240 nm (black line) [23]. After treatment with increasing molar equivalents of complex, the CD signal of c-myc DNA markedly changed. The strength of the positive signal at 265 nm decreased by 87.4% for Pu27 DNA and 71.2% for Pu22 DNA, and simultaneously an obvious negative induce CD signal at a wavelength of 295 nm appeared both for Pu27 and Pu22 DNA. These results further confirmed that the complex can interact with *c-myc* DNA tightly through groove mode, which is similar to the Pu27 binding behavior of $[(\eta^6-C_6H_6)Ru(p-$ ClPIP)Cl]Cl reported by Wu [20], and Pu22 binding properties of [Ru(bpy)2(p-{p-BEPIP BEPIP)](ClO₄)₂ = 2-(4-phenyacetylenephenyl)-1Himidazo[4,5f][1,10]phenanthroline} reported by Zhang [11].

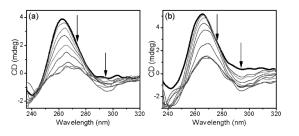


Figure 3. CD spectra of *c-myc* quadruplex DNA at increasing concentrations of ruthenium complex in Tris-KCl buffer. (a) *c-myc* Pu27 DNA, $[DNA] = 6 \mu M$, $[Ru] = 0.80 \mu M$ (b) *c-myc* Pu22 DNA, $[DNA] = 6 \mu M$, $[Ru] = 0.80 \mu M$.

For validating the binding stoichiometry of the ruthenium complex with *c-myc* quadruplex DNA, a continuous variation analysis was used. Fixed the total concentration of ruthenium complex and *c-myc* G-quadruplex DNA, the fluorescence intensities of the mixed solutions of complex and *c-myc* DNA with different concentration ratios are different. As given in **Fig. 4**, the plots of $\Delta I vs$ X showed that the points of intersection of the complex with *c-myc* DNA were found to be X = 0.5 and 0.55 for Pu27 and Pu22, respectively. The data are consistent with the 1:1 [quadruplex]/[complex] binding mode, which suggests that a specific Ru-quadruplex interaction occurs with a single guanine tetrad.

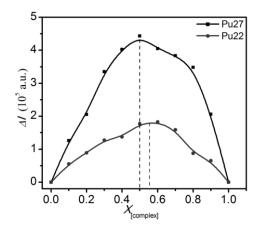


Figure 4. Job plots resulting from the continuous variation analysis for ruthenium complex with *c-myc* Pu27 and Pu22 quadruplex DNA.

3.2 Formation of *c-myc* G-quadruplex DNA induced by ruthenium complex

To evaluate the inhibitory activity of the ruthenium complex against Tag polymerase by inducing the c-myc G-quadruplex conformation, a PCR-stop assay was carried out. The sequence of c-myc DNA and its corresponding complementary sequence can hybridize to a final double-stranded DNA PCR product when used with Taq DNA polymerase as the catalyst [11]. However, in the presence of some G-quadruplex stabilizers, the template sequence *c-myc* DNA will be induced into a G-quadruplex structure that blocked the hybridization, and in this case, the final PCR product will not be detected. The effect of the complex on the hybridization of c-myc Pu27 and Pu22 DNA was illustrated in Fig. 5. It is clear that when the complex was added to the solution, the replication of *c-myc* oligomer was suppressed. The greater the concentration of the complex is, the smaller is the amount of final product that can be detected. At the complex concentration of 4.0 µM, the band of the PCR product for both Pu27 and Pu22 DNA completely disappeared, which is attributed to the Gquadruplex structure induced to block the hybridization with a complementary strand. It is noteworthy that the 4.0 µM for the complex that displays complete inhibition is much lower than 12 μ M, 15 μ M, > 40 μ M that reported for $[(bpy)_2Ru(bpibp)Ru(bpy)_2](ClO_4)_2$ [29], $[Ru(bpy)_2(dhipH_3)](ClO_4)_2$ (dhipH₃ = 3,4-dihydroxyl-imidazo[4,5-*f*][1,10]phenanthroline) [23], $[Ru(bpv)_2(p-$ TEPIP)](ClO₄)₂ [11] and [Ru(bpy)₂(p-BEPIP)](ClO₄)₂ [11], as listed in Table 1, which evidences that the complex behaves as a promising inducer of c-myc Gquadruplex DNA. However, in contrast to the efficient induction of c-myc Gquadruplex DNA, the complex could not influence human telomeric DNA (HTG21) to fold into G-quadruplex as seen from the results of CD spectra and PCR-stop assay (Fig. 6), which further implies the action selectivity of the complex towards c-myc DNA.

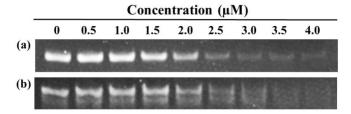


Figure 5. Effect of ruthenium complex on the hybridization of *c-myc* Pu27 (a) and Pu22 (b) DNA in PCR-stop assay.

 Table 1. The concentration of ruthenium complex for complete inhibition in PCR-stop assay.

Complex	Concentration of complete inhibition (G-quadruplex DNA)	Ref.
the present complex	4 µM (Pu27)	This work
[(bpy) ₂ Ru(bpibp)Ru(bpy) ₂](ClO ₄) ₄	12 µM (Pu27)	[29]
[(phen) ₂ Ru(bpibp)Ru(phen) ₂](ClO ₄) ₄	10 µM (Pu27)	[29]
[Ru(bpy) ₂ (dhipH ₃)](ClO ₄) ₂	15 µM (Pu27)	[23]
[Ru(phen) ₂ (dhipH ₃)](ClO ₄) ₂	12 µM (Pu27)	[23]
[(η ⁶ -HC ₆ H ₅)Ru(<i>m</i> -MOPIP)Cl]Cl	12 µM (Pu27)	[32]
[(η ⁶ -CH ₃ C ₆ H ₅)Ru(<i>m</i> -MOPIP)Cl]Cl	9 μM (Pu27)	[32]
[Ru(bpy) ₂ (<i>p</i> -TEPIP)](ClO ₄) ₂	>40 µM (Pu22)	[11]
[Ru(bpy) ₂ (<i>p</i> -BEPIP)](ClO ₄) ₂	>40 µM (Pu22)	[11]
[Ru(ip) ₃](ClO ₄) ₂ ·2H ₂ O	8 µM (Pu23)	[35]
[Ru(pip) ₃](ClO ₄) ₂ ·2H ₂ O	>8 µM (Pu23)	[35]
[Ru(IP) ₂ (PIP)](ClO ₄) ₂ ·2H ₂ O	15 μM (HTG21)	[35]
[Ru ₂ (bpy) ₄ (bip-phenol)] ⁴⁺	20 µM (HTG21)	[36]
Λ -[Ru(phen) ₂ (p-MOPIP)] ²⁺	15 μM (HTG21)	[37]
Δ -[Ru(phen) ₂ (<i>p</i> -MOPIP)] ²⁺	20 µM (HTG21)	[37]
[(bpy) ₂ Ru(ebipcH ₂)Ru(bpy) ₂](ClO ₄) ₄	10 μM (HTG22)	[38]
[(bpy) ₂ Ru(mbpibH ₂)Ru(bpy) ₂](ClO ₄) ₄	10 µM (HTG22)	[38]
[(bpy) ₂ Ru(hbpibH ₂)Ru(bpy) ₂](ClO ₄) ₄	12 μM (HTG22)	[38]

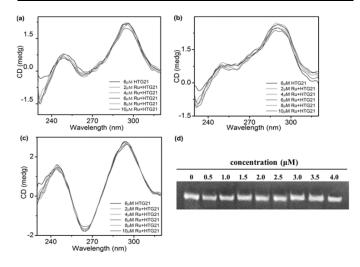


Figure 6. CD spectra of HTG21 DNA at increasing concentrations of ruthenium complex, (a) in Tris-HCl buffer, (b) in Tris-KCl buffer, (c) in Tris-NaCl buffer, (d) effect of ruthenium complex on the hybridization of HTG21 DNA in PCR-stop assay.

To further identify the existence of G-quadruplex with the naked eye, we have adopted an effective and visible method reported by Liu's group [**39**]. It is well known that G-rich DNA sequences can readily transform to G-quadruplex structures in vitro at physiological concentrations of Na⁺ and K⁺, and Gquadruplexes have the ability to bind with hemin to form the peroxidase-like DNAzymes [**39**]. It is proven that in the presence of DNAzymes, H₂O₂-mediated oxidation of TMB can be sharply accelerated and the color change is very sensitive and easy to identify. The design is based on this principle. As shown in **Fig. 7**, the H₂O₂-mediated oxidations of colorless TMB to a blue product was observed with the complex present, as well as control K⁺. This phenomenon indicates that Pu27-mer and Pu22-mer can fold into G-quadruplex under the induction of complex, and such quadruplex structure is able to bind hemin to form the hemin-G-quadruplex DNAzyme. But for the complex with double strands of ct-DNA, the solution still presents colorless. The reason is obvious, because ct-DNA cannot form G-quadruplex structure.

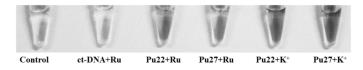


Figure 7. Characterization of the DNAzyme functions of Pu27 and Pu22 DNA in the presence of 500 nM K⁺ and 500 nM ruthenium complex in the TMB-H₂O₂ system. Conditions: TMB, 266 mM in Tris-MES bufer (25 mm MES, pH =5.10); H₂O₂, 794 mM; DNA, 500 nM; hemin, 500 nM.

3.3 Stabilization of c-myc G-quadruplex DNA by ruthenium complex

To investigate the thermodynamic stability of c-myc G-quadruplex DNA with the ruthenium complex, a FRET melting point assay was performed to detect the melting temperature (T_m) [38, 40-43]. The F27T and F22T DNA containing fluorophores at both the 5'-end and the 3'-end was used. The emission of fluorescence was normalized between 0 and 1, and the $T_{\rm m}$ was defined as the melting temperature for which the normalized emission is 0.5. The enhanced melting temperature ($\Delta T_{\rm m}$) was calculated by subtracting the $T_{\rm m}$ of the free fluorescence-labeled oligonucleotide F27T or F22T from the $T_{\rm m}$ of the nucleic acid in the presence of the complex, representing the ability of the complex to stabilize the G-quadruplex DNA. The normalized melting curves of F27T and F22T DNA treated the complex were illustrated in Fig. 8. The $T_{\rm m}$ value of free F27T in K⁺ buffer was found to be 72 °C under our experimental conditions. However, the stabilization effect of the complex on c-myc F27T DNA was proved to be relatively weak, which gave the largest $\Delta T_{\rm m}$ value of only 3.5 °C at the concentration of $[Ru] = 0.8 \mu M$. Similar result appeared for F22T, of which the T_m value was raised maximaly but only 3.0 °C at the 0.8 µM concentration of complex. Moreover, in the presence of increasing amount of double helical ct-DNA in the concentration rang of 1-5 μ M, the melting temperatures of F27T and F22T gradually decreased again, as shown in Fig. S3. The observed weak ability of the complex to stabilize c-myc quadruplex DNA may be due to the serious steric hindrance for interacting with DNA, which arises from the bulky structure of the complex. This phenomenon is also similar to [Ru2(bpy)4(bipphenol)](ClO₄)₄ {bip-phenol = 2,4-bis(1H-imidazo[4,5-f][1,10] phenanthroline-2-yl)phenol} with the greatest $\Delta T_{\rm m}$ value of 3 °C for HTG21 at concentration of $[Ru] = 0.5 \mu M$. This result would benefit from further investigation.

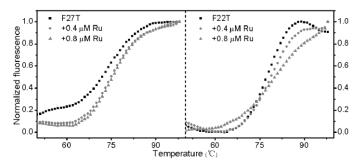


Figure 8. Normalized FRET melting curves of F27T and F22T alone, and with increasing concentrations of ruthenium complex measured by real-time PCR system.

3.4 Topoisomerase inhibition

Topoisomerase enzyme is fundamental part of the cellular processes. It has been reported that in cancerous cells, Topo I enzyme can make transient breaks in single strand of supercoiled pBR322 DNA and covalently linked to DNA to form stable Topo I–DNA complexes. The inhibitors of topo I act on Topo I–DNA complexes and finally lead to DNA damage during replication process [44-46]. Therefore, topoisomerase enzyme is considered to be a promising drug target in the chemotherapy. The results of Topo I inhibition assay by different concentrations of the ruthenium complex were given in Fig. 9. The supercoiled plasmid DNA can be entirely relaxed by Topo I in the absence of the complex (lane 2). As the concentration of the complexes increased (1-4 μ M), the amount of the relaxed DNA decreased gradually (lanes 3-6), implying that the complex may block the DNA strand passage event of the enzyme and may serve as an inhibitor of Topo I.

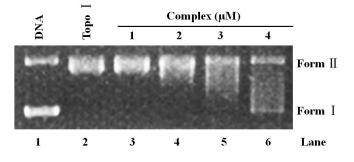


Figure 9. Effect on the activity of DNA Topo I by varying concentrations of ruthenium complex.

3.5 In vitro cytotoxicity

We evaluated the *in vitro* cytotoxic activity of the complex on four selected human cancer cell lines (A549, Huh-7, MGC-803, MCF-7) and one normal cell line (293T) using the MTT assay to determine the potential of the complex as an anticancer agent. Cisplatin was included as a positive control. The above cell lines were incubated with different concentrations of the complex for 48 h, and the IC₅₀ values for the complex against these cell lines are summarized in Table 2. As shown in Table 2, the complex displayed moderate cytotoxic activity against MCF-7 and Huh-7 and simultaneously exhibited less toxic activity toward the normal human cell 293T, while it did not show appreciable inhibition of the cell growth toward A549 and MGC-803 with IC₅₀ > 100 μ M. The previous results of cellular imaging and cytotoxicity experiments of the complex also found that it can rapidly penetrate through the HeLa cells (human cervical carcinoma cell) membrane and had somewhat antiproliferative effect on HeLa cell at the imaging concentration [31]. Therefore, the complex can be a candidate of anticancer agents, although its cytotoxic action is lower than that of cisplatin.

Table 2. The IC₅₀ values of the ruthenium complex toward different cell lines.

Compound	IC ₅₀ values (µM)					
	A549	Huh-7	MGC-803	MCF-7	293T	
the present complex	>100	67.08 ± 5.45	>100	43.46 ± 4.01	80.79 ± 5.41	
cisplatin	18.63 ± 0.26	18.27 ± 0.75	13.47 ± 0.26	12.05 ± 0.03	20.72 ± 1.23	

4. CONCLUSION

A calix[4]arene-based binuclear ruthenium(II) complex has been successfully achieved. The interaction of the complex with *c-myc* G-quadruplex DNA were studied by photophysical and biophysical means of spectroscopic titrations, CD titration spectra, color reaction studies, Job plot, PCR-stop assay, and FRET assay. The anticancer activity of the complex was evaluated by MTT assay, and the Topoisomerase inhibition was examined as well. The results suggested that the complex could bind to *c-myc* DNA by good affinity, with the Pu27 and Pu22 binding constants as large as 1.18×10^7 and 4.13×10^6 M⁻¹, respectively. It was found to interact with *c-myc* G-quadruplex DNA in groove mode with a 1:1 binding ratio, and notably it was identified to be an efficient inducer of *c-myc* G-quadruplex DNA although with weak stabilization effect on *c-myc* G-quadruplex DNA.

The complex also displayed TopoI inhibition activity and antiproliferative activity toward MCF-7 and Huh-7 tumor cells, with IC₅₀ values of (67.08 ± 5.45) μ M and (43.46 ± 4.01) μ M for MCF-7 and Huh-7 lower than (80.79 ± 5.41) μ M for 293T normal cells. In sum, the complex can be developed as a potential inhibitor in chemotherapy through its interaction with *c-myc* G-quadruplex DNA.

ACKNOWLEDGEMENTS

The authors thank the National Natural Science Foundation (No. 21401038), Natural Science Foundation of Hebei Province (No. B2016201122), the Science and Technology Development Plan Project of Baoding City (No.14ZF009), the Science and Technology Development Plan Project in Tai'an City (No. 2015GX2048), and the Scientific Research Foundation of Taishan University (No. Y-01-2014017).

CONFLICT OF INTEREST

On behalf of all authors, the corresponding author states that there is no conflict of interest.

REFERENCES

- E. Babu, J. Bhuvaneswari, P.M. Mareeswaran, P. Thanasekaran, H.M. Lee, S. Rajagopal, Coordin. Chem. Rev. 380, 519–549, (2019).
- G. Piraux, L. Bar, M. Abraham, T. Lavergne, H. Jamet, J. Dejeu, L. Marcélis, E. Defrancq, B. Elias, Chem. Eur. J. 23, 11872–11880, (2017).
- J. Rubio-Magnieto, S. Kajouj, F.D. Meo, M. Fossépré, P. Trouillas, P. Norman, M. Linares, C. Moucheron, M. Surin, Chem. Eur. J. 24, 15577– 15588, (2018).
- J. Weynand, A. Diman, M. Abraham, L. Marcélis, H. Jamet, A. Decottignies, J. Dejeu, E. Defrancq, B. Elias, Chem. Eur. J. 24, 19216–19227, (2018).
- A.R. Simović, R. Masnikosa, I. Bratsos, E. Alessio, Coordin. Chem. Rev. 398, 113011–113036, (2019).
- 6. V. Brabec, J. Kasparkova, Coord. Chem. Rev. 376, 75–94, (2018).
- 7. M. Pal, U. Nandi, D. Mukherjee, Eur. J. Med. Chem. 150, 419-445, (2018).
- J.P. Liu, Z. Chen, T.W. Rees, L.B. Ke, L.N. Ji, H. Chao, Coord. Chem. Rev. 363, 17–28, (2018).
- G.L. Liao, X. Chen, J.H. Wu, C. Qian, Y. Wang, L.N. Ji, H. Chao, Dalton Trans. 44, 15145–15156, (2015).
- X. Chen, J.H. Wu, Y.W. Lai, R. Zhao, H. Chao, L.N. Ji, Dalton Trans. 42, 4386–4397, (2013).
- S.Y. Zhang, Q. Wu, H. Zhang, Q. Wang, X.C. Wang, W.J. Mei, X.H. Wu, W.J. Zheng, J. Inorg. Biochem. 176, 113–122, (2017).
- H.J. Yu, Y. Zhao, W.J. Mo, Z.F. Hao, L. Yu, Spectrochim. Acta. A: Mol. Biomol. Spectrosc. 132, 84–90, (2014).
- A.P. Saraswati, N. Relitti, M. Brindisi, S. Gemma, D. Zisterer, S. Butini, G. Campiani, Drug Discov. Today 24, 1370–1388, (2019).
- R. Catalano, F. Moraca, J. Amato, C. Cristofari, R. Rigo, L.D. Via, R. Rocca, A. Lupia, A. Maruca, G. Costa, B. Catalanotti, A. Artese, B. Pagano, A. Randazzo, C. Sissi, E. Novellino, S. Alcaro, Eur. J. Med. Chem. 182, 111627, (2019).
- 15. N. Kim, Curr. Med. Chem. 26, 2898-2917, (2019).
- B. Rajasekhar, C. Kumar, G. Premkumar, M.A. Bin Riyaz, P.T.V. Lakshmi, T. Swu, Struct. Chem. 30, 727–742, (2019).
- X.N. Wang, X.X. Su, S.Q. Cheng, Z.Y. Sun, Z.S. Huang, T.M. Ou, Expert Opin. Ther. Pat. 29, 353–367, (2019).
- 18. M.I. Umar, D.Y. Ji, C.Y. Chan, C.K. Kwok, Molecules 24, 2416, (2019).
- K.B. Wang, M.S.A. Elsayed, G.H. Wu, N.J. Deng, M. Cushman, D.Z. Yang, J. Am. Chem. Soc. 141, 11059–11070, (2019).
- Q. Wu, K.D. Zheng, S.Y. Liao, Y. Ding, Y.Q. Li, W.J. Mei, Organometallics 35, 317–326, (2016).
- L. Li, H.M. Liu, X.K. Liu, S.Y. Liao, Y.T. Lan, Q. Wu, X.C. Wang, Q. Wang, S.Y. Zhang, W.J. Mei, RSC Adv. 7, 23727–23734, (2017).
- Z. Zhang, Q. Wu, X.H. Wu, F.Y. Sun, L.M. Chen, J.C. Chen, S.L. Yang, W.J. Mei, Eur. J. Med. Chem. 80, 316–324, (2014).
- Y. Liu, Y.N. Liu, L.C. Yang, C.W. Cao, Y.H. Zhou, J. Liu, Med. Chem. Commun. 5, 1724–1728, (2014).
- 24. Z. Zhang, X.H. Wu, F.Q. Sun, F. Shan, J.C. Chen, L.M. Chen, Y.S. Zhou, W.J. Mei, Inorg. Chim. Acta 418, 23–29, (2014).
- Z. Zhang, W.J. Mei, X.H. Wu, X.C. Wang, B.G. Wang, S.D. Chen, J. Coord. Chem. 68, 1465–1475, (2015).
- L. Scaglioni, R. Mondelli, R. Artali, F.R. Sirtori, S. Mazzini, BBA-Gen. Subjects 1860, 1129–1138, (2016).

- T. Lemarteleura, D. Gomeza, R. Paterskia, E. Mandineb, P. Maillietb, J. Rioua, Biochem. Bioph. Res. Co. 323, 802–808, (2004).
- C. Rajput, R. Rutkaite, L. Swanson, I. Haq, J.A. Thomas, Chem. Eur. J. 12, 4611–4619, (2006).
- C.P. Zheng, Y.N. Liu, Y. Liu, X.Y. Qin, Y.H. Zhou, J. Liu, J. Inorg. Biochem. 156, 122–132, (2016).
- K. Suntharalingam, A.J.P. White, R. Vilar, Inorg. Chem. 49, 8371–8380, (2010).
- Q.Y. Huang, Y.F. Han, Z.B. Zheng, Chinese J. Inorg. Chem. 34, 217-229, (2018).
- 32. Q. Wu, T.F. Chen, Z. Zhang, S.Y. Liao, X.H. Wu, J. Wu, W.J. Mei, Y.H. Chen, W.L. Wu, L.L. Zeng, W.J. Zheng, Dalton Trans. 43, 9216–9225, (2014).
- A. Ghosh, P. Das, M.R. Gill, P. Kar, M.G. Walker, J.A. Thomas, A. Das, Chem. Eur. J. 17, 2089–2098, (2011).
- S. Mardanya, S. Karmakar, D. Mondal, S. Baitalik, Inorg. Chem. 55, 3475– 3489, (2016).
- J.N. Zhang, Q.Q. Yu, Q. Li, L.C. Yang, L.M. Chen, Y.H. Zhou, J. Liu, J. Inorg. Biochem. 134, 1–11, (2014).
- H.Q. Zhao, X.X. Xu, S. Wang, Y.X. Mi, Z.B. Zheng, X.L. Zhao, Transit. Metal Chem. 43, 539–548, (2018).

- D.D. Sun, Y.N. Liu, D. Liu, R. Zhang, X.C. Yang, J. Liu, Chem. Eur. J. 18, 4285–4295, (2012).
- L. Xu, X. Chen, J.H. Wu, J.Q. Wang, L.N. Ji, H. Chao, Chem. Eur. J. 21, 4008–4020, (2015).
- Q. Li, J.N. Zhang, L.C. Yang, Q.Q. Yu, Q.C. Chen, X.Y. Qin, F.L. Le, Q.L. Zhang, J. Liu, J. Inorg. Biochem. 130, 122–129, (2014).
- S.N. Georgiades, N.H. Abd Karim, K. Suntharalingam, R. Vilar, Angew. Chem. Int. Ed. 49, 4020–4034, (2010).
- W. Streciwilk, A. Terenzi, X.L. Cheng, L. Hager, Y. Dabiri, P. Prochnow, J.E. Bandow, S. Wölfl, B.K. Keppler, I. Ott, Eur. J. Med. Chem. 156, 148– 161, (2018).
- Q. Wu, S.Y. Liao, G.N. Yu, J. Wu, W.J. Mei, J. Inorg. Biochem. 189, 81– 90, (2018).
- K. Duskova, S. Sierra, M.S. Arias-Pérez, L. Gude, Bioorg. Med. Chem. 24, 33–41, (2016).
- R. Gaur, M. Usman, Spectrochim. Acta. A: Mol. Biomol. Spectrosc. 209, 100-108, (2019).
- 45. G.L. Ma, X.D. Bi, F. Gao, Z. Feng, D.C. Zhao, F.G. Lin, R. Yan, D.D. Liu, P. Liu, J.B. Chen, H.B. Zhang, J. Inorg. Biochem. 185, 1-9, (2018).
- Y.C. Wang, C. Qian, Z.L. Peng, X.J. Hou, L.L. Wang, H. Chao, L.N. Ji, J. Inorg. Biochem. 130, 15–27, (2014).