

POTENTIAL OF *CURATELLA AMARICANA* L. AGAINST SARS-COV2: BIOAVAILABILITY, MOLECULAR SIMILARITY AND MOLECULAR DOCKING BETWEEN SECONDARY METABOLITES AND PROTEASE TYPE 3-CHYMOTRYPSIN (3CLPRO)

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ABSTRACT

We report the bioavailability analysis for six secondary metabolites with antiviral, antioxidant and antitumor activity reported, from *Curatella amaricana* L. Additionally, the molecular similarity analysis of each metabolite is presented and compared with Lopinavir, Ritonavir, Darunavir, Cobicistat and Nelfinavir, which actually are in the third phase for the production of a new vaccine for SARS-Cov2. The mode of interaction through molecular docking between each structure and the zone of action for protease type 3-chymotrypsin (3CLpro) also is presented. The molecular geometry for structures were optimized at semiempirical PM6 level. The bioavailability and molecular docking calculations were performed using the algorithms incorporated in chemoinformatic servers and AutoDock Vina. The results show that the structures studied lead a moderated permeability through the cell membrane, by complying with Lipinski's "rule of 5". Molecular similarity was evaluated by averaging geometric parameters (3D-Shape) and electrostatic potential (ESP). The results show that the most secondary metabolites would have a similar mode of action as the Lopinavir, with average similarity between 0.65 and 0.73. This last idea is reinforced by the results for molecular docking with the 3CLpro active site, highlighting the interaction of the molecules studied with the amino acid residues: His-41, Phe-140, Gly-143, Ser-144, Cys-145, His-163, Glu-166 and His-172, with an range interaction-free energy between -7.2 kcal/mol and -9.2 Kcal/mol, highlighting Quercetin 3-O-Alpha-L-rhamnoside with improve affinity energy than Lopinavir.

Keywords: SARS-Cov2, secondary metabolites, *Curatella amaricana* L., protease type 3-chymotrypsin (3CLpro).

INTRODUCTION

Since the emergence of a new strain of coronavirus called SARS-CoV-2, which causes pneumonia and bronchiolitis, known as covid-19, originating from Wuhan, Hubei, China, a variety of candidate compounds have been reported for the treatment of this virus. According to a recent report, the compounds Lopinavir, Ritonavir, Darunavir, Cobicistat and Nelfinavir are in the third phase for the production of a new vaccine. These compounds have in common their action on protease type 3-chymotrypsin (3CLpro) [1]. 3CLpro together with PLpro (papain-like protease), correspond to the two main proteases in virus replication. These enzymes cleave the two polyproteins PP1A and PP1AB with individual functional components. Due to its role in virus replication, 3CLpro has become a drug target [2].

In order to explore new sources of possible compounds against SARS-CoV-2, we present the bioavailability analysis and molecular docking for six secondary metabolites with antitumor and antiprotease potential, which can be extracted from the plant diversity of our country, Venezuela [3]. Specifically, we have chosen *Curatella amaricana* L., a species in the Dilleniaceae family, known in Venezuela as "Chaparro", "Chaparro sabanero" or "Curata". It is a representative pyrophilic tree of the plain and hot land savannas of Venezuela (Amazonas, Aragua, Bolívar, Carabobo, Delta Amacuro, Federal District, Falcón, Lara Miranda, Trujillo and Zulia). This species is recognized by its low and often twisted habit, leathery leaves, thick reddish bark, white flowers and unpleasant smells [4]. With our results, we hope to give a molecular insight to the experimental findings on the use of secondary metabolites of *C. amaricana* L. in the treatment of SARS-CoV-2.

COMPUTATIONAL DETAILS

Geometric Optimization and Molecular Similarity analysis

Molecular similarity principle states that molecules with similar structure tend to have similar properties. Indeed, the observation that common substructural fragments lead to similar biological activities, can be quantified from database analysis [5], [6]. The secondary metabolites studied are shown in Figure 1. These compounds are flavonoids derived from Quercetin, whose antiprotease activity is well documented [7]. For the similarity analysis, we used as references the structures of Lopinavir, Ritonavir, Darunavir, Cobicistat and Nelfinavir, which are known for their interaction with 3CLpro.

For all structures, the geometric optimizations were carried out at semiempirical PM6 level using GAMESS software package [8]. The molecular similarity calculations were carried out with ShaEP software package.

ShaEP performs rigid-body superimposition of 3D molecular models, using a matching algorithm [9]. Two characteristic scores were calculated for comparison: 3D shape and electrostatic potential (ESP) [10]. These scores range is from 0 to 1, in which 0 and 1 correspond to no similarity and the same molecules, respectively.

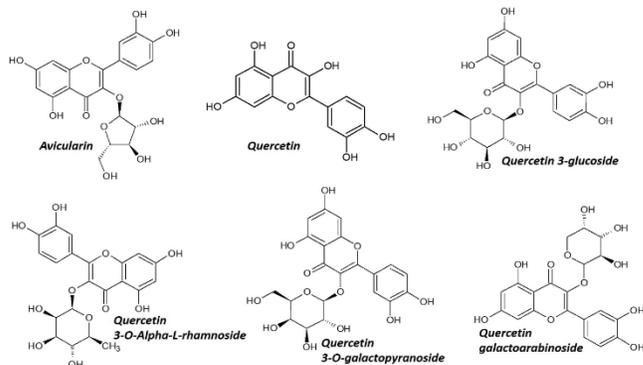


Figure 1. Secondary metabolites of *Curatella amaricana* L. ligands studied.

Bioavailability prediction

Properties of molecules such as bioavailability or membrane permeability have often been connected to simple molecular descriptors such as logP (partition coefficient), molecular weight (MW), or counts of hydrogen bond acceptors and donors in molecule [11]. These descriptors are included in the named Lipinski "Rule of Five" [12]. The rule states that most molecules with good membrane permeability have logP ≤ 5, molecular weight ≤ 500, number of hydrogen bond acceptors ≤ 10 and number of hydrogen bond donors ≤ 5. In order to evaluate the bioavailability of the secondary metabolites studied, the Lipinski's parameters were calculated using Molinspiration Cheminformatics software [13].

Preparation of main protease and molecular docking

Structure of the 3CLpro at 1.82 Å (PDB ID: 2GTB) of resolution was used for studied the interaction with the secondary metabolites of *C. amaricana* L. [14]. From PDB structure we retained an 3CLpro complex consisting of only A chain (302 residues, from Gly-2 to Cys-300). Hydrogen atoms were added to the model using VEGA ZZ package [15]. Atomic charges were assigned using the

RESULTS AND DISCUSSION

Bioavailability (Lipinski parameters)

Lipinski's rule is widely used to determine molecular properties that are important for drug's pharmacokinetic in vivo. Table 1 contains calculated percentage of absorption (%ABS), molecular polar surface area (TPSA) and Lipinski parameters of the secondary metabolites investigated.

Gasteiger-Marsili method [16]. Aza-peptide epoxide and others crystal molecules were removed. Secondary metabolites ligands were optimized using semiempirical PM6 level using GAMESS software package. Later, the ligands were prepared for docking experiment defining rotatable bonds using AutoDock Tools version 1.5.6. A grid box size of 50, 60, 50 Å was generated and allocated at the center of the receptor binding site using x, y and z coordinates of 8.876, -4.039 and -1.654. Molecular docking simulations, interactions calculations and visualization of all structures were performed using AutoDock Vina and VEGA ZZ softwares. Co-crystallized ligands were re-docked as validation of the docking protocol.

Table 1. Calculated percentage of absorption (%ABS), molecular polar surface area (TPSA) and Lipinski parameters of the *Curatella americana* L. secondary metabolites investigated.

Molecule	%ABS	Volume	TPSA	Nrot	H acceptors	H donors	Log P	MW
Avicularin	43.4	347.36	190.28	4	11	7	0.80	434.35
Quercetin	63.7	240.08	131.35	1	7	5	1.68	302.24
Quercetin 3-O-galactopyranoside	36.4	372.21	210.50	4	12	8	-0.36	464.38
Quercetin galactoarabinoside	43.5	347.36	190.28	3	11	7	0.06	434.35
Quercetin 3-glucoside	36.4	372.21	210.50	4	12	8	-0.36	464.38
Quercetin 3-O-Alpha-L-rhamnoside	43.4	363.95	190.28	3	11	7	0.64	448.38

Molecular hydrophobicity or lipophilicity is indicated by octanol/water partition coefficient (Log P). Hydrophilic/lipophilic nature of drug molecule affects drug permeability across cell membrane. Log P values of all the secondary metabolites studied were found to be lower than 5, in agreement with Lipinski's rule of five. Quercetin 3-O-galactopyranoside and Quercetin 3-glucoside presents the lowest Log P (-0.36), indicating that this metabolites are of hydrophilic character, suggesting poor permeability across cell membrane. However, Quercetin and similar flavonoids has been reported for its antiviral activity [17]. In fact, the Quercetin action against SARS-CoV across 3CLpro interaction ($IC_{50} = 23.8 \mu M$) also has been reported [18]. The rest of the metabolites studied have Log P values between 0.64 and 1.68, indicating greater lipophilicity and therefore better permeability through the cell membrane.

Total polar surface area (TPSA) is closely related to the hydrogen bonding potential of a molecule and is a good predictor of drug transport properties such as intestinal absorption, bioavailability and blood brain barrier penetration. Molecules with a polar surface area of greater than 140 \AA^2 tend to be poor at permeating cell membranes [19]. For molecules to penetrate the blood-brain barrier a PSA less than 90 \AA^2 is usually needed [20]. TPSA for Quercetin lead a value of 131 \AA^2 , which in agreement with the limits mentioned above. The rest of the structures studied do not agreement with this range of values. Number of rotatable bonds is a simple topological parameter that measures molecular flexibility and is considered to be a good descriptor of oral bioavailability of drugs. It has been shown that higher oral bioavailability is associated with lower rotatable bond count. Rotational bonds make the compounds flexible, hence easily interact with specific rigid binding area [21]. All the structures studied show low-moderate molecular flexibility due to range of rotatable bonds (1-5). Number of hydrogen bond acceptors (O and N atoms) and number of hydrogen

bond donors (NH and OH) in the tested compounds were found to be not within Lipinski's limit i.e. less than 10 and 5 respectively. The percentages of absorption for title compounds calculated from TPSA ranged between 43% and 63%, indicated a moderate oral bioavailability. When the BOILED-Egg model [22] was applied through SwissADME server [23], we found a good permeation blood-brain barrier (BBB) probability for Quercetin.

Molecular Similarity

The aim of this paper is the exploration of the antiviral potential against SARS-CoV of the secondary metabolites of *C. americana* L., through the interaction with 3CLpro. The combination of the literature review and the search for similar structures related to *C. americana* L. secondary metabolites in the PubChem and DrugBank databases, allowed to choose five compounds: Lopinavir, Ritonavir, Darunavir, Cobicistat and Nelfinavir. According to a recent report, these compounds are in the third phase for the production of a new vaccine. These compounds have in common their action on 3CLpro [1]. The mechanisms of actions of these drugs are clarified and some molecular targets are validated, and available from the Protein Data Bank. Starting from the knowledge of five molecules identified as antiviral drugs we analyzed the molecular similarity between secondary metabolites of *C. americana* L. Two important properties, 3D-shape and electrostatic potential (ESP) of secondary metabolites were compared to those drugs and the results are shown in Table 2. It is shown that secondary metabolites studied has a high shape similarity with all reference molecules, with average from of 0.63 (Cobicistat) to 0.78 (Nelfinavir). These molecules possess a similar framework of rings, mainly substituted by -OH groups. It is also shown that the five reference drugs have a high ESP similarity with the most secondary metabolites studied, with average values from 0.67 to 0.73.

Table 2. 3D-shape and electrostatic potential (ESP) of the *Curatella americana* L. secondary metabolites investigated

Molecule/Reference	Cobicistat		Darunavir		Lopinavir		Nelfinavir		Ritonavir	
	shape	ESP	shape	ESP	shape	ESP	shape	ESP	shape	ESP
Avicularin	0.65	0.77	0.73	0.71	0.71	0.75	0.79	0.71	0.69	0.77
Quercetin	0.57	0.76	0.72	0.69	0.67	0.68	0.71	0.75	0.58	0.70
Quercetin-3-O-galactopyranoside	0.67	0.74	0.73	0.64	0.74	0.75	0.81	0.75	0.72	0.75
Quercetin galactoarabinoside	0.61	0.69	0.72	0.67	0.73	0.73	0.78	0.69	0.70	0.66
Quercetin 3-glucoside	0.62	0.70	0.75	0.62	0.75	0.72	0.81	0.69	0.72	0.75
Quercetin 3-O-Alpha-L-rhamnoside	0.65	0.74	0.71	0.67	0.74	0.77	0.80	0.63	0.70	0.76

When both, 3D-shape and ESP values are averaged, we observed a high similarity of the metabolites studied with Lopinavir. Lopinavir has been reported in experiments of molecular dynamic as a powerful inhibitors of coronavirus

protease [24]. In fact, an Arbidol and Liponavir-Ritonavir combination is currently being used in the preclinical testing phase [25]. The molecular similarity of the metabolites studied with Liponavir and the other reference

molecules, suggest a possible mode of action in the molecular targets of interest for SARS-Cov2.

Molecular Docking

In the present study, molecular docking was performed to identify the docking score of six structures of the Figure 1 towards of the 3CLpro at 1.82 Å (PDB ID: 2GTB). Figure 2 and Figure 3 shows the principal interaction mode and affinity energy in the binding site of 3CL protease for the secondary metabolites studied. His-41, Phe-140, Gly-143, Ser-144, Cys-145, His-163, Glu-166 and His-172 are the main aminoacids in the pocket of 3CLpro, in agreement with the crystallographic structure reported [14].

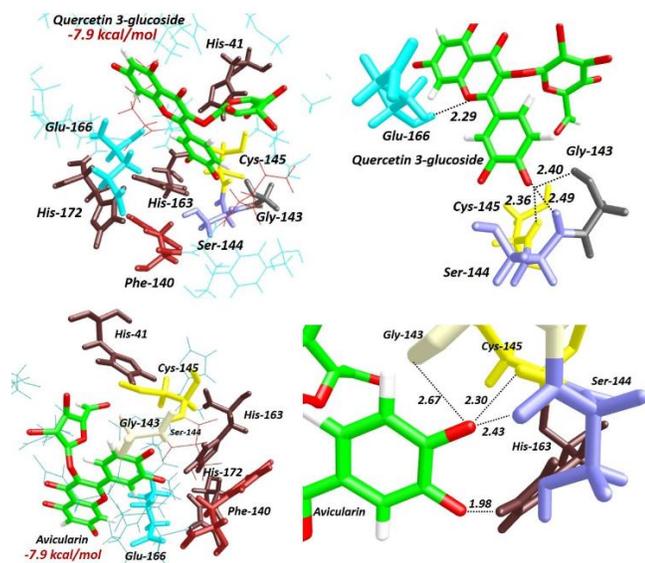


Figure 2. Interaction mode for secondary metabolites ligands in the 3CLpro binding site. Hydrogen bond distances are showed in Å. Affinity energy are showed in kcal/mol.

The high affinity of drug compounds depends on the type and amount of bonding that occurs with the active site of the protein. Both, Avicularin and Quercetin 3-glucoside molecules have the same affinity energy (-7.9 kcal/mol). Likewise, both molecules coincide in three of the four hydrogen bonds in the active site of the 3CLpro (Gly-143, Ser-144, Cys-145). However, the stabilization energy due to hydrogen bonding is different, being -1.21 kcal/mol for avicularin and -2.12 kcal/mol for Quercetin 3-glucoside. This difference can be attributed to the position of the fourth hydrogen bond. For Avicularin this bond is established with His-163 residue and for Quercetin 3-glucoside this bond is formed with Glu-166 residue.

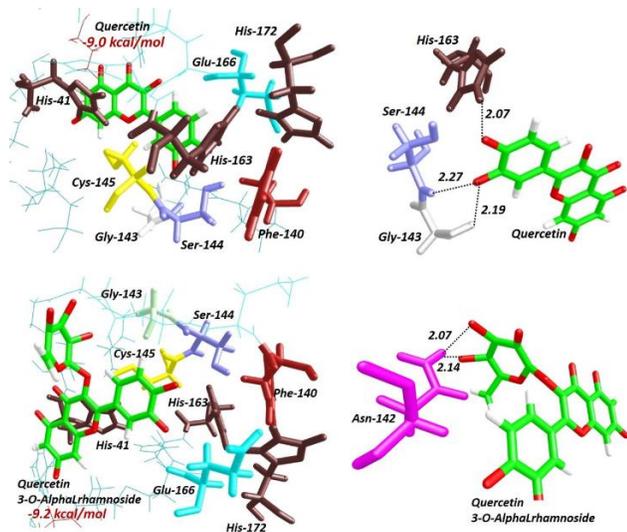


Figure 3. Interaction mode for secondary metabolites ligands in the 3CLpro binding site. Hydrogen bond distances are showed in Å. Affinity energy are showed in kcal/mol.

Likewise, the stabilization energy by hydrophobic interaction in Quercetin 3-glucoside is greater (-6.14 kcal/mol, His-41/Met-49/Cys-145/His-163/Met-165) than Avicularin (-5.44 kcal/mol, His-41/Met-49/Cys-145/His-163) due to additional interaction with Met-165.

Quercetin and Quercetin 3-O-Alpha-L-rhamnoside shows the higher affinity energy values, -9.0 kcal/mol and -9.2 kcal/mol, respectively (Figure 3). Both molecules interacting in the active site of the 3CLpro with His-41, Phe-140, Gly-143, Ser-144, Cys-145, His-163, Glu-166 and His-172. The stabilization energy due to hydrogen bonding is different, being -1.79 kcal/mol for Quercetin 3-O-Alpha-L-rhamnoside (Asn-142) and -2.68 kcal/mol for Quercetin (Gly-143/Ser-144/His-163). Regarding hydrophobic interactions, the stabilization energy values are similar, being -3.25 kcal/mol for Quercetin 3-O-Alpha-L-rhamnoside and -3.38 kcal/mol for Quercetin. As mentioned earlier in this article, the potential of flavonoids as possible agents against SARS-Cov2 has been widely reported [26]. In order to compare with the reference compounds and their action with 3CLpro, the Table 3 show the affinity energy for the all the compounds.

Table 3. Energy affinity (kcal/mol) for de secondary metabolites studied and reference compounds.

Molecule	Affinity energy
Avicularin	-7,9
Quercetin	-9,0
Quercetin-3-O-galactopyranoside	-7,8
Quercetin galactoarabinoside	-8,7
Quercetin 3-glucoside	-7,9
Quercetin 3-O-Alpha-L-rhamnoside	-9,2
Cobiciclat	-8,1
Darunavir	-7,8
Lopinavir	-8,5
Nelfinavir	-9,9
Ritonavir	-7,0

Quercetin and its derivate Quercetin 3-O-Alpha-L-rhamnoside shows the best energy affinity, in accordance with the energy affinity Nelfinavir. Has been reported that the HIV-1 protease inhibitor, nelfinavir, strongly inhibite replication of the SARS coronavirus (SARS-CoV). Nelfinavir inhibite the cytopathic effect induced by SARS-CoV infection and the expression of viral antigens is much lower in infected cells treated with nelfinavir than in untreated infected cells [27]. On the other hand, Quercetin galactoarabinoside values for energy affinity are comparable with the energy affinity for both, Cobiciclat and Lopinavir. According to a recent report, Lopinavir and Cobiciclat are in the third phase for the production of a new vaccine against SARS-CoV2 infection [28]; [1]. According to our results, the significant molecular similarity and bioavailability found in the secondary metabolites derived from quercetin in *C. americana* L., this specie is projected as a source of extraction for new compounds in the fight against SARS-Cov2.

CONCLUSIONS

Curatella americana L. is a specie found in Latin America with a strong presence in the plain and hot land savannas of Venezuelan. Theoretical and experimental reports on the activity against SARS-Cov2 of flavonoids found in other plant species around the world, suggest *C. americana* as a possible source for these compounds. In this framework of ideas, we have determined the molecular similarity and bioavailability of the secondary metabolites derived from quercetin found in this specie. Our results show significant similarity with some antiviral reference compounds such as Lopinavir and Nelfinavir, which have currently been used against SARS-Cov2. The bioavailability of the studied

compounds is in moderate agreement with the Lipinski "Rule of Five". On the other hand, molecular docking at the 3CLpro active binding site show affinity

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DECLARATION OF INTERESTS

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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