SYNTHESIS OF BENZOXAZEPINE DERIVATIVES FROM PYRAZOLE-CHALCONE VIA A SIMPLE AND CONVENIENT PROTOCOL USING BASIC ALUMINA AS SOLID SUPPORT

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ABSTRACT

In the present study a series of novel benzoxazepine (5a-h) derivatives were synthesized by the thermal cyclization reaction of various pyrazole-chalcones (3a-h) with 2-aminophenol, by conventional heating and microwave irradiation (180 W) in solvent-free conditions in short reaction times (9–12 min), giving high yields of products (80–88%). The homogeneity of all the newly synthesized compounds has been checked by TLC. Their IR, NMR, ESI-mass spectral data and elemental analysis are in accord with the assigned structure. The title compounds were evaluated for their antibacterial activity against Gram-positive bacteria *Staphylococcus aureus, Bacillus subtilis*, and Gram-negative bacteria *Pseudomonas aeruginosa, Escherichia coli*. Compounds **3c**, **3h**, **5b**, **5c**, **5g** and **5h** were found to show good antibacterial activity when compared with that of standard drug Ampicillin. Furthermore, the same library of compounds were evaluated for antifungal activity against *Aspergillus nigerzeae*, *Penicillium italicum* and *Fusarium oxysporum* using Grieseofulvin as standard drug. The results of the above studies show that the compounds **3b**, **3c**, **3e**, **3h**, **5a**, **5c**, **5g** and **5h** showed good antifungal activity against.

Keywords: Pyrazole, Chalcone, benzoxazepine, microwave irradiation, basic alumina.

INTRODUCTION

The prominence in attaining chalcones has numerous applications in the field of pharmaceutics and also in the production of pesticides, cosmetics^{1,2} Chalcone and its functionalized derivatives display diverse pharmacological properties have been discovered as, antimicrobial³⁻⁸, antitubercular^{3,0}, anticoncer^{10,11}, anti-inflammatory¹⁰, and antioxidant⁷ activities. Chalcones bear very good synthon framework so that variety of novel heterocycles with good pharmaceutical profile can be designed. On the other hand, pyrazoles represent a key motif in heterocyclic chemistry and occupy a prime place in medicine and pesticide chemistry due to their capability to exhibit a wide range of bioactivities such as antimicrobial, anti-inflammatory, anticancer and antiviral¹²⁻¹⁵. Pyrazoles were found to possess inhibitory activities against XO, cyclooxygenase, and alkaline phosphatases^{16, 17}. Benzoxazepine derivatives are prevalent designs of biologically active compounds^{18,19}. These derivatives are well-known pharmacophores showing capable properties against various diseases such as antipsychotic²⁰, central nervous system damage²¹ and anticancer activities22. Usually, benzoxazepine derivatives preparation involves multistep synthesis²³⁻²⁵. Therefore, an efficient, eco-friendly, straightforward synthetic method toward benzoxazepine still represents an interesting assignment for chemists

The science of green chemistry was developed to chance the increasing demand for environmentally gentle chemical practices. Solvent free synthesis has received considerable attention in recent years. Microwave heating has a prominence in the search for green synthesis²⁶. Microwave irradiation conditions extensively reduce reaction time without promoting any side reactions^{27,28}. Alumina like solid support, holds excellent capability to absorb the organic compounds on their surface, and transmits microwave irradiation. Previous reports have also shown that the basic oxides, like alumina, assist as a good alternative for catalyzed heteroannulation reactions due to its character as a base ion in the solid framework²⁹. This provoked us to discover an efficient yet simple and one-pot green protocol using basic alumina as solid support in microwave irradiation.

In the current study, we aimed to obtain new compounds containing both pyrazole and benzoxazepine rings in the same structure, *via* the cyclocondensation reaction, solvent free microwave irradiation conditions. This nonconventional protocol offers several applications such as simple procedure, fast reaction conditions and good yields as compared to conventional methods. It involves the exposure of neat reactants to give high yields of pure products, easy work-up, low cost and economical. The use of basic alumina as a solid support assists and it eliminates the use of K_2CO_3 and ethanol solvent. Herein, we have shown (**Scheme 1**) the synthesis of benzoxazepines derivatives from pyrazole-chalcone reaction with 2-amino phenol, basic alumina as a solid support using as a catalyst under microwave irradiation (180 W). In all reactions the products were achieved in high yield. Chemical structures of compounds were confirmed by IR, ¹H NMR, ¹³C NMR, ESI-MS, spectral data and elemental analysis.

EXPERIMENTAL

All the melting points were determined in open capillary tubes and are uncorrected. The purity of the compounds was checked by TLC on silica gel 60 F_{254} (Merck). ¹H NMR and ¹³C NMR spectra were recorded on Avance 400 spectrometer using CDCl₃ solvent TMS as an internal standard. IR spectra were recorded in KBr Shimadzu FTIR 8400 S spectrophotometer. Mass spectra were recorded on LCMS-2010A Shimadzu spectrophotometer. Elemental analysis was determined by using the EA1112 Thermofinnigan CHNS analyzer. Microwave reactions were carried out in a Multisynth series microwave system (Milestone).

Conventional heating method (A)

A mixture of chalcone (**3a-h**) [30] (0.01mol), 2-amino phenol (7) (0.01 mol) were dissolved in 10 ml of ethanol and to this K_2CO_3 (2 g) was added. The contents were refluxed for appropriate time mentioned in **Table-1**. The progress of the reaction was monitored by TLC. After completion of the reaction, the mixture was cooled to room temperature and poured into water. The product was extracted with acetone and was recrystalized with acetone to afford pure compounds (**5a-h**).

Microwave irradiation method (B)

A mixture of chalcone (**3a-h**) (0.01mol), 2-amino phenol (7) (0.01 mol) was adsorbed on basic alumina (3 g). The adsorbed material was taken in a quartz tube and inserted into the Teflon vial with screw capped and then subjected to microwave irradiation for an appropriate given time in **Table-1**. After completion of the reaction, cooled to room temperature and the product was extracted with acetone and was recrystalized with acetone to afford pure compounds (**5a-h**).

2-((E)-2,3-dihydro-2-(1,3-diphenyl-1H-pyrazol-4-yl)benzo[b][1,4] oxazepin-4-yl)phenol (5a)

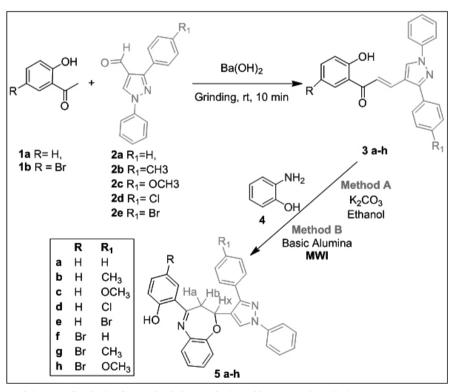
White solid; IR spectrum, ν cm⁻¹: 3135 (OH), 1607 (C=N), 1238 (C-O-C). ¹HNMR spectrum, δ , ppm: 8.12 (s, 1H, pyrazole-H), 7.98-6.99 (m, 18H, Ar-H), 5.67 (dd, 1H, J_{XB} =12.4H₂, J_{XA} =3.2H₂), 3.24 (dd, 1H, J_{BA} =16.8H_z, J_{BX} =12.4H₂), 3.04 (dd, 1H, J_{AB} =16.8H_z, J_{AX} =3.2H₂). ¹³CNMR spectrum, δ_{C} , ppm: 154.5, 152.1, 147.3, 139.8, 133, 129.5, 128.7, 128.6, 128.3, 128.2, 126.7, 126.1, 125.7, 124.9, 124.7, 124, 123.2, 120.3, 119.5, 119.2, 112.5, 70, 42.5. Found, %: C 78.71; H 5.04; N 9.23. C₃₀H₂₃N₃O₂. Calculated, %: C 78.75; H 5.07; N 9.18. MS: 458 [M+H]⁺.

Compound Number	Molecular Formula (Mol. Wt)	M.P. (°C)	Conventional heating (A)		Microwave irradiation (B)	
		M.r. (C)	Time (h)	Yield (%)	Time (min)	Yield (%)
5a	$C_{30}H_{23}N_{3}O_{2}$ (457)	180-182	6	60	9	85
5b	$C_{31}H_{25}N_{3}O_{2}$ (471)	193-195	8	62	10	83
5c	$C_{31}H_{25}N_{3}O_{3}$ (487)	188-190	8	70	10	85
5d	$C_{30}H_{22}CIN_{3}O_{2}$ (491)	176-178	8	68	12	80
5e	$C_{30}H_{22}BrN_{3}O_{2}$ (535)	222-224	9	65	9	88
5f	$C_{30}H_{22}BrN_{3}O_{2}(535)$	254-256	9	62	9	85
5g	$C_{31}H_{24}BrN_{3}O_{2}(549)$	226-268	7	60	10	80
5h	$C_{31}H_{24}BrN_{3}O_{3}(565)$	252-254	8	62	9	82

Table 1: Conventional heating and Microwave irradiation optimization for the synthesis of 5a-h.

2-((E)-2,3-dihydro-2-(1-phenyl-3-p-tolyl-1H-pyrazol-4-yl)benzo[b][1,4] oxazepin-4-yl)phenol (5b)

White solid; IR spectrum, ν cm⁻¹: 3141 (OH), 1611 (C=N), 1231 (C-O-C). ¹HNMR spectrum, δ , ppm: 8.13 (s, 1H, pyrazole-H), 7.98-7.07 (m,17H, Ar-H), 5.67 (dd, 1H, J_{XB} =12.4H₂, J_{XA} =2.8H₂), 3.22 (dd, 1H, J_{BA} =16.8H₂, J_{BX} =12.4H₂), 3.01 (dd, 1H, J_{AB} =16.8 H₂ J_{AX} =2.8 H₂), 2.41 (s, 3H, CH₃). ¹³CNMR spectrum, $\delta_{\rm C^9}$ ppm: 161.4, 152.2, 140.3, 138.4, 136.1, 129.5, 129.4, 128.5, 127.3, 126.8, 121.8, 121.6, 119.7, 119.6, 119.3, 118.2, 72.4, 43.8, 21.8. Found, %: C 78.93, H 5.32, N 8.94. C₃₁H₂₅N₃O₂. Calculated, %: C 78.96; H 5.34; N 8.91. MS: 472 [M+H]⁺.



Scheme 1: Synthesis of pyrazole-chalcones (3a-h) and benzoxazepines (5a-h)

2-((E)-2,3-dihydro-2-(3-(4-methoxyphenyl)-1-phenyl-1H-pyrazol-4-yl) benzo[b][1,4]oxazepin-4-yl)phenol (5c)

White solid; IR spectrum, $v \text{ cm}^{-1}$: 3142 (OH), 1600 (C=N), 1239 (C-O-C). ¹HNMR spectrum, δ , ppm: 8.11 (s, 1H, pyrazole-H), 7.98-6.98 (m,17H, Ar-H), 5.65 (dd, 1H, J_{XB} =12.4H₂, J_{XA} =2.8H₂), 3.86 (s, 3H, OCH₃), 3.23 (dd, 1H, J_{BA} =16.8H₂, J_{BX} =12 H₂), 3.01 (dd, 1H, J_{AB} =16.8H₂, J_{AX} =2.8H₂). ¹³CNMR spectrum, δ_{c} , ppm: 163.7, 161.2, 160.1, 153.7, 151.8, 140.1, 135.9, 129.6, 129.3, 127.1, 126.6, 125.4, 121.6, 119.4, 119.2, 118, 114.3, 72.3, 55.3, 43.5, 29.5. Found, %: C 76.40, H 5.18, N 8.67. C₃₁H₂₅N₃O₃. Calculated, %: C 76.37; H 5.17; N 8.62. MS: 488 [M+H]⁺.

2-((E)-2-(3-(4-chlorophenyl)-1-phenyl-1H-pyrazol-4-yl)-2,3dihydrobenzo[b][1,4]oxazepin-4-yl) phenol (5d) White solid; IR spectrum, v cm⁻¹: 3141 (OH), 1602 (C=N), 1229 (C-O-C).

White solid; IR spectrum, v cm⁻¹: 3141 (OH), 1602 (C=N), 1229 (C-O-C). ¹HNMR spectrum, δ , ppm: 8.13 (s, 1H, pyrazole-H), 7.99-7.06 (m, 17H, Ar-H), 5.65 (dd, 1H, J_{xB} =12H₂, J_{xA} =3.2H₂) 3.26 (dd, 1H, J_{BA} =14.2, J_{BX} =12H₂), 3.03 (dd, 1H, J_{AB} =14.2H₂, J_{AX} =3.2H₂). ¹³CNMR spectrum, δ_{C} , ppm: 160.9, 150.8, 139.8, 136.1, 134.6, 131.3, 129.6. 129.4, 128.8, 127,1, 126.9, 121.8, 121.2, 119.4, 119.3, 117.9, 71.9, 43.3. Found, %: C 73.21; H 4.48; N 8.57. C₃₀H₂₇ClN₃O, Calculated, %: C 73.24; H 4.51; N 8.54. MS: 492 [M+H]⁺.

2-((E)-2-(3-(4-bromophenyl)-1-phenyl-1H-pyrazol-4-yl)-2,3dihydrobenzo[b][1,4]oxazepin-4-yl) phenol (5e)

White solid; IR spectrum, $v \text{ cm}^{-1}$: 3140 (OH), 1604 (C=N), 1225 (C-O-C). ¹HNMR spectrum, δ , ppm: 8.13 (s, 1H, pyrazole-H), 7.99-7.05 (m, 17H, Ar-H), 5.66 (dd, 1H, $J_{XB} = 12.4H_Z$, $J_{XA} = 3.2H_Z$), 3.23 (dd, 1H, $J_{BA} = 16.8H_Z$, $J_{BX} = 12.4H_Z$), 3.03 (dd, 1H, $J_{AB} = 16.8H_Z$, $J_{AX} = 3.2H_Z$). ¹³CNMR spectrum, δ_{CZ} ppm: 160.9, 150.9, 139.8, 136.1, 131.8, 129.9, 129.4, 127.1, 127, 122.7, 121.8, 121.2, 119.4, 119.3, 118, 71.9, 43.3, 29.6. Found, %: C 67.51; H 3.57; N 7.86. C₃₀H₂₂BrN₃O₂. Calculated, %: C 67.17; H 4.13; N 7.83. MS: 535 [M]⁺, 537 [M+2]⁺.

4-bromo-2-((E)-2,3-dihydro-2-(1,3-diphenyl-1H-pyrazol-4-yl)benzo[b] [1,4]oxazepin-4-yl)phenol (5f)

White solid; IR spectrum, $v \text{ cm}^{-1}$: 3402 (OH), 1600 (C=N), 1238 (C-O-C). ¹HNMR spectrum, δ , ppm: 8.09 (s, 1H, Pyrazole-H), 7.78-6.97 (m, 17H, Ar-H), 5.64 (dd, 1H, J_{XB} =12H₂, J_{XA} =3.2H₂), 3.21 (dd, 1H, J_{BA} =16.8H₂, J_{BX} =12H₂), 3.02 (dd, 1H, J_{AB} =16.8H₂, J_{AX} =3.2H₂). ¹³CNMR spectrum, δ_{C} , ppm: 160, 158, 154.6, 151, 141.1, 130.4, 129.4, 129.3, 127.5, 126.4, 125.8, 125.5, 122.3, 119, 116.7, 114.3, 70.1, 40.88. Found, %: C 67.14; H 4.11; N 7.86. C₃₀H₂₂BrN₃O₂. Calculated, %: C 67.17; H 4.13; N 7.83. MS: 535 [M]⁺, 537 [M+2]⁺.

4-bromo-2-((E)-2,3-dihydro-2-(1-phenyl-3-p-tolyl-1H-pyrazol-4-yl) benzo[b][1,4]oxazepin-4-yl) phenol (5g)

White solid; IR spectrum, ν cm⁻¹: 3137 (OH), 1602 (C=N), 1210 (C-O-C). ¹HNMR spectrum, δ, ppm: 8.14 (s, 1H, Pyrazole-H), 7.81-6.98 (m, 16H, Ar-H), 5.64 (dd, 1H, J_{XB} =12.4H₂, J_{XA} =2.8H₂), 3.21 (dd, 1H, J_{BA} =16.8H₂, J_{BX} =12.4H₂), 2.99(dd, 1H, J_{AB} =16.8H₂, J_{AX} =2.8H₂), 2.36 (s, 3H, CH₃). ¹³CNMR spectrum, δ_{C} , ppm: 159.2, 151.9, 140, 137.1, 132.8, 131.2, 129.3, 128.5, 128.4, 128.3, 126.7, 126.6, 120.9, 119.7, 119.3, 117.8, 72.1, 43.6, 20.2. Found, %: C 67.61; H 4.35; N 7.67. C₃₁H₂₄BrN₃O₂. Calculated, %: C 67.64; H 4.39; N 7.63. MS: 549 [M]⁺, 551 [M+2]⁺.

4-bromo-2-((E)-2,3-dihydro-2-(3-(4-methoxyphenyl)-1-phenyl-1Hpyrazol-4-yl)benzo [b] [1,4] oxazepin-4-yl)phenol (5h)

White solid; IR spectrum, ν cm⁻¹: 3135 (OH), 1598 (C=N), 1209 (C-O-C). ¹HNMR spectrum, δ, ppm: 8.09 (s, 1H, Pyrazole-H) 8.08-6.93 (m, 16H, Ar-H), 5.64 (dd, 1H, J_{xB} =11.6H₂, J_{xA} =3.2H₂), 3.86 (s, 3H, OCH₃), 3.21 (dd, 1H, J_{BA} =16.8H₂, J_{BX} =11.6H₂), 3.02 (dd, 1H, J_{AB} =16.8H₂, J_{AX} =3.2H₂). ¹³CNMR spectrum, δ_{c} , ppm: 152.96, 150.87, 147.5, 139.69, 134.33, 133.69, 131.58, 129.82, 129.53, 128.84, 126.91, 126.11, 125.05, 124.30, 122.67, 119.19, 112.85, 70.3, 43.4, 29.7. Found, %: C 65.70; H 4.25; N 7.46. C₃H₂₄BrN₃O₃. Calculated, %: C 65.73; H 4.27; N 7.42. MS: 565 [M]⁺, 567 [M+2]⁺.

BIOLOGICAL ASSAY

Antibacterial activity

The synthesized novel compounds **3a-h**, **5a-g** were screened for their Antibacterial activity against different types of bacterial strains, they are Gram positive bacterial strains of *Bacillus subtilis and Staphylococcus aeureus*, Gram negative bacterial strains of *Pseudomonas aeruginosa* and *Escherichia coli*, at a concentration of 50 µg/mL. The cultures were diluted with 5% autoclaved saline and the final volume was made with concentration approximately 10^5 - 10^6 CFU/mL. The synthesized compounds were diluted in acetone for antibacterial biological assays. For agar disc diffusion method³⁰, the liquid form of test compound was soaked on to the disc and then allowed to air dry, such that the disc gets completely saturated with test compound. The saturated chemical discs were introduced onto the upper layer of the medium evenly loaded with the bacteria.

The discs were dipped in different chemical samples, were placed over the evenly spread bacterial nutrient media and incubated at 37 °C for 24 to 48 h for better inhibition of bacteria. The zones of inhibition were measured after 24 to 48 h. All the experiments were carried out in triplicates and the results were expressed as zone of Inhibition in mm. The zones of inhibition of synthesized compounds **3a-h**, **5a-h** were compared with the zone of inhibition of standard antibiotic concentration of Ampicillin (50 μ g/mL). The Antibacterial activity was evaluated and the results are presented in **Table 2**.

	Zone of Inhibition, mm					
Compound	Gram-positive	bacteria	Gram-negative bacteria			
compound	Staphylococcus aureus	Bacillus subtilis	Pseudomonas aeruginosa	Escherichia coli		
3a	23	05	04	10		
3b	24	10	08	25		
3c	30	12	09	32		
3d	22	07	08	25		
3e	12	04	03	12		
3f	10	08	02	23		
3g	22	10	07	22		
3h	28	12	11	29		
5a	22	08	05	17		
5b	27	10	08	28		
5c	28	11	08	32		
5d	20	07	05	22		
5e	18	10	08	21		
5f	15	09	06	18		
5g	28	11	09	28		
5h	32	11	10	28		
Ampicillin	30	12	10	30		

Table 2: Antibacterial activity of compounds 3a-h and 5a-h.

Antifungal activity

The antifungal activity of synthesized compounds **3a-h**, **4a-h** was tested against three pathogenic fungi, namely *Aspergillus niger*, *Penicillium italicum*, and *Fusarium oxysporum*, by the poison plate technique³¹ at a concentration of 50 µg/mL. Three kinds of fungi were incubated in PDA at 25 ± 1 °C for 5 days to get new mycelium for antifungal assay, then a mycelia as disks of approximately 0.45 cm diameter cut from the culture medium were picked up with a sterilized inoculation needle and inoculated in the center of PDA plate. Test compounds were dissolved in acetone (10 mL) then added to the Potato Dextrose Agar medium (PDA, 90 mL). The final concentration of compounds in the medium was adjusted to 50 µg/mL. The inoculated plates were incubated at 25 ± 1 °C for 5 days. Acetone was diluted with sterilized distilled water and used as control, while Grieseofulvin (50 µg/mL) was used as standard control for each treatment three replicates of experiments were carried out. The radial growth of the fungal colonies was measured on the sixth day. The Antifungal activity was evaluated and the results are presented in **Table 3**.

RESULTS AND DISCUSSION

Chemistry

Several synthetic methodologies are available for the synthesis of chalcones. Generally the condensation of acetophenone with aldehyde in the presence of basic media is widely used. By keeping the view in green methods we have selected to synthesize an efficient solvent free synthesis of chalcones by grinding equimolar quantities of 2-hydroxy acetophenone and substituted pyrazole aldehyde in the presence of Ba(OH),, at room temperature for 10 min. Further, these pyrazole-chalcones (**3a-h**)^{32,33} on thermal cyclization with 2-aminophenol in the presence of basic alumina under microwave irradiation^{34,35} resulted benzoxazepines (5a-h) in good yields (Scheme 1). All the reactions were carried out using microwave irradiations in 9-12 min., while same reactions under conventional condition, the time period required for the completion of the reaction was quite long and gave moderate yields. Hence, the microwave assisted synthesis bids clean and cheaper alternative path to that of conventional heating, indicating that the microwave irradiation simplifies the polarization of molecules producing reaction to occur at shorter reaction times in good yields. The use of microwave irradiation in present investigation gave 80-88% yields of product, whereas moderate yields were obtained by conventional methods. All the synthesized compounds were characterized by ¹H NMR, ¹³C NMR, Mass and IR spectral data.

	Zone of Inhibition, mm					
Compound	Aspergillus nigerzeae	Penicillium italicum	Fusarium oxysporum			
3a	08	16	22			
3b	12	21	25			
3c	11	20	23			
3d	09	20	26			
3e	12	24	28			
3f	09	21	23			
3g	06	18	23			
3h	11	21	24			
5a	15	25	28			
5b	08	14	19			
5c	14	21	24			
5d	07	15	18			
5e	06	12	14			
5f	08	10	10			
5g	14	21	23			
5h	13	22	24			
Grieseofulvin	12	20	25			

Table 3: Antifungal activity of Compounds 3a-h and 5a-h.

Biological activities

Antibacterial activity

All the compounds were screened for their antibacterial activity against *Staphylococcus aureus*, *Bacillus subtilis*, *Pseudomonas aeruginosa* and *Escherichia coli* using Ampicillin as standard drug. The activity was determined using cup plate agar diffusion method by measuring the zone of inhibition in mm (Table 2). The compounds were screened at the concentration of 50 μ g/ml in DMSO. From the screening studies it is evident that the synthesized compounds 3c, 3h, 5b, 5c, 5g and 5h showed good antibacterial activity against all the tested organisms, which shows that the electron releasing groups enhanced the antibacterial activity. Furthermore, methoxy (3c), bromo methoxy (3h), methyl (5b), methoxy (5c), bromo methoxy (5h) group derivatives are potentially active.

Antifungal activity:

All the compounds were screened for their antifungal activity against *Aspergillus nigerzeae*, *Penicillium italicum* and *Fusarium oxysporum* using Grieseofulvin as standard drug. The activity was determined using cup plate agar diffusion method by measuring the zone of inhibition in mm. The compounds were screened at the concentration of 50 μ g/ml in DMSO. From the screening studies it is evident that the synthesized compounds **3b**, **3c**, **3e**, **3h**, **5a**, **5c**, **5g** and **5h** showed good antifungal activity against all the tested organisms. However, in the case of the remaining compounds (**Table 3**).

CONCLUSIONS

In summary, we have demonstrated an efficient, economical, environmentally benign, and rapid process for the synthesis of benzoxazepines derivatives from pyrazole-chalcones reaction with 2-amino phenol. It was discovered that microwave-assisted method is highly efficient procedure for the preparation of benzoxazepines, especially in the solvent-free media. This method eliminates the use of K_2CO_3 and ethanol solvent. No additional base and solvent are required as basic alumina acts as solid support and base. Additionally, application of microwave decreases the time of the reaction extensively with a better-quality yield of the product. The uniqueness of the methodology lies in its eco-friendly operation, with excellent yield. In this study, a new hybrid molecules consisting of biologically important benzoxazepine derivatives (5a-h) from pyrazole-chalcones (3a-h) pharmacophores were synthesized and their antibacterial and antifungal activities were determined. Amongst the synthesized compounds, **3c**, **3h**, **5b**, **5c**, **5g** and **5h** showed good antibacterial activity. The antifungal activity of the synthesized compounds also showed that compounds **3b**, **3c**, **3e**, **3h**, **5a**, **5c**, **5g** and **5h** have good inhibition values. The experimental results of this study will likely provide a new basis for the design of interesting pyrazole-chalcones based benzoxazepine, and further studies, including the design of new analogs of the heterocyclic moiety, are underway.

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